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(54) Drug administration.

(57) Compositions and methods useful for the prevention or treatment of a human or animal disorder or for the regulation of a human or animal physiological condition are provided. The compositions used comprise, in admixture, a biologically-effective amount of a drug specific for the disorder or condition and a biocompatible, water-soluble, amphiphilic steroid, other than a natural bile salt, which is capable of increasing drug permeability of the human or animal body surface across which the drug is to be administered, in an amount effective to increase the permeability of the surface to the drug.

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DRUG ADMINISTRATION

Part of the work described and claimed herein was supported in part by a grant or award from the United States Government, which has certain rights in the invention.

INTRODUCTION

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This invention relates to the administration of drugs across human or animal body surfaces. (As used herein, the term "drug" is defined as any biologically-active chemical or natural substance useful for treating a medical or veterinary disorder, preventing a medical or veterinary disorder, or regulating the physiology of a human being or animal.) More particularly, the invention relates to an administration method based on the use of biocompatible, water-soluble, amphiphilic steroids capable of increasing the permeability of human and animal body surfaces to a variety of biologically-active substances. Application of admixtures of steroid and drug to mucosal or epithelial surfaces advantageously results in enhanced drug delivery across the body surface.

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25BACKGROUND OF THE INVENTION

To elicit its characteristic biological response in the body, a drug must be available in an effective concentration at its site of action. The concentration of a drug that reaches its reactive site varies with such factors as the amount of drug administered, the extent and

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rate of its absorption, distribution, binding or
localization in tissues, its biotransformation, and its
excretion. (For a review of these topics, see Goodman and
5 Gilman's, The Pharmacological Basis of Therapeutics, 6th
edition, MacMillan Publishing Co., Inc., New York, 1980,
pp. 1-39.) The foregoing factors, and hence the ultimate
efficacy of a particular drug, are in turn influenced by
the route chosen for drug administration.

10 The common routes of drug administration are
enteral (oral ingestion) and parenteral (intravenous,
subcutaneous, and intramuscular) routes of
administration. To determine the appropriate mode of drug
15 administration, it is necessary to understand some of the
advantages and disadvantages of the route used. For
example, intravenous drug administration is advantageous
for emergency use when very rapid increases in blood
levels are necessary. The intravenous route allows for
dosage adjustments when required, and is also useful for
20 administration of large volumes of a drug when diluted.
However, there are limitations on the usefulness of
intravenous drug administration. One problem is the risk
of adverse effects resulting from the rapid accumulation
of high concentrations of the drug in plasma and tissues.
25 Consequently, intravenously administered drug solutions
must generally be continuously monitored and injected
slowly. The intravenous route is not suitable for oily or
insoluble substances. Furthermore, intravenous
administration is restricted to trained medical personnel.
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Other routes of parental administration are often
inconvenient or painful for patients especially if
frequent administration is required. Subcutaneous
35 injection is used for drugs that are not irritating. This
mode of administration is not suitable for delivering

larger volumes are suitable for administering irritating substances which may cause pain or necrosis at the site of injection. Intramuscular administration is generally suitable for moderate volumes, oily substances, and some irritating substances. The intramuscular route cannot be used during anticoagulant medication and may interfere with the interpretation of certain diagnostic tests.

Oral administration of drugs is generally more convenient and economical and is most acceptable to humans. However, this route of administration requires patient cooperation. Absorption may be inefficient (i.e., incomplete) for poorly soluble, slowly absorbed, or unstable drug preparations, and the time from ingestion to absorption may prohibit effective use in emergency situations. Furthermore, peptides and proteins will often be destroyed by the digestive enzymes, acid, and surface-active lipids in the gut prior to reaching the site of action.

Certain drugs which need to be administered frequently are not effectively absorbed when administered orally and hence must be delivered by injection methods. Yet, a number of problems are associated with conventional injection therapies.

By way of illustration, conventional insulin therapy requires frequent insulin injections resulting in discomfort and disruption of the patient's lifestyle. Hence, many diabetics either refuse insulin therapy altogether or avoid intensive treatment regimens such as those which involve injections with each meal. In addition, certain patients, especially young children, elderly patients, and those who are blind and/or disabled,

oft n hav difficulty with insulin self-administrati n by
injection. Furthermore, insulin absorpton after
subcutaneous injection is variable in terms of rate and
5 amount depending upon factors such as exercise, local
blood flow, depth and volume of injection, the presence of
local proteases which degrade insulin, and perhaps other,
unknown factors. Even presently available short acting
and long acting preparations of insulin or mixtures
10 thereof cannot mimic the daily glucose and insulin
excursions of non-diabetic individuals. Portable infusion
pumps have now been employed to increase the ease of
delivering subcutaneously meal-related insulin boluses.
However, these devices are externally worn and are
15 therefore cumbersome. They require regular needle
replacement, are complicated by local infections at the
site of needle placement, are expensive, and are not
acceptable to many patients.

20 It is clear that a reproducible, reliable, and
non-invasive means for delivering drugs such as insulin
would be highly desirable. What is needed especially in
the case of insulin is a delivery system that would permit
easy, rapid, and non-invasive administration of insulin at
25 meal times when blood glucose concentration rises to peak
levels. Since the discovery of insulin six decades ago,
there have been many attempts to develop alternate means
of insulin delivery. For instance, insulin has been
administered enterally, either alone or encapsulated in
30 liposomes (microcapsules), sublingually, vaginally, and
rectally, with and without surfactants.

In addition to the preceding routes of
administration, the nasal route has been the subject of
35 investigation for the delivery n t only f insulin but f
ther drugs as well. It is known that certain very small

peptides can be absorbed through the nasal mucosa as a "snuff" or directly from aqueous solution without an adjuvant. Examples of peptides which can sometimes be administered by this route are adrenocorticotrophic hormone (ACTH), luteinizing hormone releasing hormone (LHRH), oxytocin and vasopressin. Indeed, for patients with diabetes insipidus, the intranasal route is frequently the means for vasopressin delivery.

In contrast to these directly administrable compounds, many drugs such as insulin are inefficiently absorbed across mucous membranes at physiological pH in the absence of adjuvants. Several workers have attempted to mix insulin with adjuvants that might enhance nasal insulin absorption. Hirai et al., Int. J. Pharmaceutics (1981) 9:165-184; Hirai et al., Diabetes (1978) 27:296-299; British Pat. No. 1,527,605; and U.S. Pat. No. 4,153,689; and Pontiroli et al. (1982) Br. Med. J. 284:303-386, have described the use of various bile salts to enhance absorption of insulin by the nasal mucosa.

While the nasal mucosal route has received considerable attention for systemic drug delivery, it has also hitherto been known that drugs may be applied to mucous membranes of the conjunctiva, nasopharynx, oropharynx, ear canal, respiratory tract, vagina, rectum, colon, and urinary bladder for their local effects.

SUMMARY OF THE INVENTION

We have discovered an effective means of administering a drug to a human being or animal which avoids many of the problems associated with other modes of administration such as injection. The invention provides methods and compositions useful for the prevention and/or

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wherein each dashed line, independently, represents a single or a double bond; D is a group which renders an effective amount of the steroid water soluble within the range of about pH 2 to about pH 12; E and G are individually OAc, OH, or a lower alkyl or lower heteroalkyl; W is OAc or H; and Q, V and X are individually OH or H. Most preferably the steroids of formula (1) have functional groups as follows: E is β OAc, α OH, or a lower (3 or fewer carbons) alkyl or heteroalkyl group in β position; G is α OAc, OH, lower alkyl, or lower heteroalkyl; W is α OAc or H; Q is OH or H, provided that, when W is α OAc and Q is α OH, Q must be α -axial; V is H or α OH; and X is H or α OH; provided that the steroid contains two or three polar functions exclusive of D. Three OH groups are allowed provided that one is C₁₆ α -axial, replacing OAc at that position. (As used herein, the symbol "OAc" refers to the acetoxyl radical OCOCH₃.)

The steroid of formula (1) can be unconjugated, i.e., D is O⁻Na⁺, O⁻K⁺, O⁻Rb⁺, O⁻Cs⁺, or some other ionic configuration, or it can be conjugated, i.e., D is an organic group containing at least one carbon atom. Preferably group D has a molecular weight below about 600 daltons and is one of the following groups:

(a) a peptide of one, two, or three amino acids and containing an ionic function which is dissociated within the range of about pH 2 to about pH 12;

(b) a heteroalkyl group of about three or fewer carbon atoms which contains an ionic function which is dissociated within the range of about pH 2 to about pH 12;

(c) a uric acid of about six or fewer carbon atoms which contains an ionic function which is dissociated within the range of about pH 2 to about pH 12;

(d) a polyether containing between about six and about fourteen carbon atoms, inclusive, which terminates in an ionic function which is dissociated within the range of about pH 2 to about pH 12; or

(e) a polyether containing between about sixteen and about twenty-four carbon atoms, inclusive, and optionally terminating in an ionic function which is dissociated within the range of about pH 2 to about pH 12.

Group D is preferably bonded to C₂₁ via an amide or ester linkage.

Preferably the steroid used in the invention is characterized in that the unconjugated derivative of the steroid is retained on a hydrophobic column for a length of time sufficient to produce a k' factor value of at least about 4, the k' factor value being obtained by subjecting a monomeric solution of 1 mg/ml of such steroid derivative to high-performance liquid column chromatography at 3,000 psi, using a 250 x 4.6 mm column having octadecylsilane-coated 5 µm silica particles as the stationary phase and a mobile phase, delivered at 1.0 ml/min., consisting of 75% methanol in water, v/v, buffered with 0.005 M KH₂PO₄/H₃PO₄ to give an apparent pH value, as measured using a glass electrode, of 5.0, the k' factor value being defined by $k' = \frac{t_r - t_0}{t_0}$,

where t₀ is the retention time in the column of the solvent front and t_r is the retention time in the column of the steroid derivative as measured by obtaining the

elution profile of the steroid derivative by absorbance at 210 nm.

5 Preferably the steroid is further characterized
in that the critical micellar temperature (CMT) (the
temperature at which the steroid ceases to be an insoluble
crystal or gel and begins to go into solution and
self-associate in solution) of an aqueous 1% solution,
10 w/v, of the steroid is below human or animal body
temperature, and optimally below about 0°C within the
range of about pH 2 to about pH 12 (a measure of
solubility); and the critical micellar concentration (CMC)
(the concentration at which the steroid ceases to be an
15 ideal solution and begins to self-associate) is as high as
8 mMolar but preferably less than 4 mM and more preferably
less than 2 mMolar at 37°C in 0.15 M NaCl as measured by
surface tension.

20 Preferred steroids are ionized or partially
ionized alkali salts of fusidic acid, 24,25-
dihydrofusidic acid, cephalosporin P₁ and C₂₁
conjugates of these; and tauro-24,25-dihydrofusidate and
glyco-24,25-dihydrofusidate. Other preferred steroids may
25 be 17,20-24,25-tetrahydrofusidic acid, 3-acetoxyl-fusidic
acid, cephalosporin P₂-P₅, and C₂₁ conjugates of
these; and tauro-17,20-24,25-tetrahydrofusidate, tauro-
16α-OH-24,25-dihydrofusidate, tauro-16α-OH-17,20-24,25-
tetrahydrofusidate, tauro-16-O-methyl-ether-24,25-
30 dihydrofusidate, tauro-16-O-methyl-ether-17,20-24,25-
tetrahydrofusidate. The foregoing preferred steroids must
be freely soluble in water at the pH of the composition to
be administered. A broad spectrum of drugs may be used
including but not limited to (a) peptides and polypeptides
35 which have a molecular weight between about 100 and about
300,000 daltons, (b) non-peptides, and (c) other drugs.

The invention permits more effective, safer and convenient administration across a body surface of a human being or animal, e.g., any mucosal surface, including but not limited to oropharynx, ear canal, respiratory tract, nasopharynx, conjunctiva, rectal, gastric, intestinal, endometrial, cervical, vaginal, colon, urethra, urinary bladder, and tympanic membrane, of a wide range of drugs, some of which normally cannot be so administered. For example, in contrast to intravenous injection, the invention allows for convenient patient self-administration of drugs, making it more likely that the patient will adhere to prescribed treatment schedules. The invention will deliver drugs more quickly than oral, subcutaneous, or intramuscular administration and is potentially more effective in delivering drugs to a localized site than is intravenous administration. In contrast to oral administration, drugs administered by the method of the present invention do not need to pass through the liver where they might be metabolized in order to reach the site of action, in some cases reducing liver toxicity. Because drugs can be delivered more efficiently to the site of action or into the blood stream, in many cases a higher percentage of a given drug will reach the site than when administered by other means, thereby reducing the quantity of the drug that needs to be administered. Administration of reduced quantities of drug can be more cost effective than administration of the same drug by a conventional method. Alternatively, this method provides a means of delivering increased amounts of drug to a site, if desired or necessary. In addition, administration, according to the present invention, does not have deleterious toxic side effects. In particular, the steroids of the invention are able to potentiate the transport and enhance absorption of drugs across mucosal surfaces into the circulation while causing less irritation, burning sensation or other local

toxicity than is a characteristic by-product of transport facilitated by the carrier molecules.

5 The invention also allows drug administration to be tailored much more closely to cyclic disease states than is possible with other forms of administration. This is of particular importance with diseases (such as diabetes) in which drug requirements vary during the course of a day.

10 It is contemplated that the adjuvants and methods of the instant invention can be used to administer agents useful for vaccination (both active immunization with antigens and immunogenic fragments thereof as well as
15 passive immunization with antibodies and neutralizing fragments thereof) and to administer agents useful for birth control.

20 It is further contemplated that the adjuvants and methods of the instant invention may potentially be useful for the delivery across plant surfaces of antiviral agents, systemic insecticides and herbicides; and across arthropod surfaces of contact insecticides and miticides.

25 Other features and advantages of the invention will be apparent from the following description of the preferred embodiments thereof, and from the claims.

30 DESCRIPTION OF THE INVENTION

 The steroid which is admixed with a drug to be administered is preferably an ionized or partially ionized, water-soluble derivative of fusidic acid or cephalosporin P₁-P₅, preferably a derivative having
35 formula (1), above. These steroid molecules are all characterized in that they have the specific fur-ring

structure of fusidic acid and cephal sporin P₁-P₅,
including the boat conformation of the B ring (in contrast
to cholesterol derivatives such as bile salts, which have
the B ring in the lower energy, more stable chair
conformation).

The structure of the steroid molecule affects its
chemical properties and thus its functioning as a drug
transporting molecule. We believe that all of the steroid
molecules used in the invention facilitate transport by
self-associating to form reversed micelles within the
membrane across which the drug is being transported; these
reversed micelles, it is believed, function as pores,
allowing the drug to pass through. A measure of a given
steroid molecule's ability to form such reversed micelles
is the hydrophobicity of the unconjugated form of the
molecule, a property which can be quantified using the k'
factor value, which is computed by observing the steroid's
retention time in a high-performance liquid chromatography
(HPLC) column under the conditions described above. As
mentioned above, the k' value of the unconjugated
derivative of the steroid should be at least about 4 for
the steroid to be suitable in the therapeutic compositions
of the invention.

Critical micellar temperature (CMT) is an
additional measure of a steroid's utility in the
compositions of the invention. CMT is the temperature at
which the steroid molecules abruptly become soluble and
self-associate into micelles from the gel or crystalline
state. This change is a reflection of the colligative
properties of the system, and the micelles formed at a
temperature just above the CMT can be small, e.g. dimers.
The steroid molecules used in the invention should have a
great enough tendency to self-associate to give a CMT at

r below human or animal body temperature and optimally below about 0°C, for a 1% aqueous solution, w/v, within the range of about pH 2 to about pH 12.

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The steroids of formula (1) can be conjugated or unconjugated at C₂₁. The conjugating group can be any organic group which does not raise the critical micellar temperature of a 1% solution of the steroid above human or animal body temperature and preferably does not raise the CMT above about 0°C within the range of about pH 2 to about pH 12 and does not raise the CMC above about 8 mMolar at 37°C in 0.15 M NaCl as measured by surface tension. Preferably, the CMC is less than 2 mMolar under similar conditions.

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The conjugating group can be, e.g., any ionic function-containing straight or branch-chained amino acid. The amino acid can be aliphatic or aromatic, and can also be a homo- or a di-homo-amino acid, e.g. homotaurine or homoglycine, or an amphoteric amino acid, e.g., sulfobetaine or phosphobetaine. A straight or branched chain di- or tripeptide which terminates in an ionic function which is dissociated within the range of about pH 2 to about pH 12 can also be employed. Peptides larger than tripeptides generally should not be used because they can lead to unacceptably lower solubility. Any suitable uronic acid, e.g. glucuronic acid, can also be used.

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Preferred conjugating amino acids are glycine and taurine. Preferred straight-chain peptides are diglycine and glutathione, and preferred branched chain peptides are sarcosylcysteine, hydroxyprolinetaurine, and sarcosyltaurine.

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When the conjugating group is a polyether of at least sixteen carbon atoms, the group need not (although it may) contain an ionic function; the ionic function is unnecessary because such groups are highly polar and thus confer solubility without ionization. For smaller polyether groups, an ionic function is generally necessary, although it can be weakly ionizable since the smaller polyethers are polar themselves.

The group bonded to each of C_6 and C_{16} , independently, (W and E in formula (1)) can be OAc ($OCOCH_3$) as in naturally occurring fusidic acid and cephalosporin P_1 . Alternatively, E can be OH, an alkyl (e.g., methyl or ethyl) or a different heteroalkyl (e.g. alkyloxy, alkylthio, or ether derivative) group of three or fewer carbon atoms; larger groups can unacceptably lower solubility. Group G, bonded to C_3 , can be OH, as in naturally occurring fusidic acid and cephalosporin P_1 - P_5 . G can also be OAc, a lower alkyl group, or a different lower heteroalkyl group. Group W, if OAc, preferably should be in the β -axial orientation.

The molecule should possess two or three polar functions, exclusive of any side chains at C_{21} , at the positions indicated above where acetoxyl and hydroxyl groups can be located.

The k' and CMT of the steroids used in the compositions of the invention are influenced by whether the steroid is conjugated at C_{21} and, if so, by the nature of the conjugating group. Additionally, a polar group at position 16 is essential for solubility (See position E on formula 1). Because the k' factor value is influenced by the polarity of any conjugating group, unconjugated derivatives must be used in numerical

comparisons involving steroids which are conjugated with different groups, and comparisons involving both conjugated and unconjugated steroids. Overall hydrophobicity and k' factor value generally decrease as the polarity of the conjugating group increases. However, such decrease is not a reflection of the hydrophobicity of the steroid nucleus. It is this hydrophobicity which is the important parameter for purposes of reversed micelle formation.

As discussed supra, the k' factor value of the unconjugated derivative of any such steroid should be at least about 4. For example some k' values of some unconjugated steroids useful in the invention are: cephalosporin P_1 ($k'=9.5$); fusidic acid ($k'=20.7$); 3-acetoxyl-fusidic acid ($k'=26.4$); 24,25-dihydrofusidic acid ($k'=27.1$). To give a few minimum k' values of conjugated steroids, the k' factor value of a glycine-conjugated steroid should be at least about 2 to be useful in the invention. For a taurine-conjugated steroid, the k' factor value should be at least about 1.

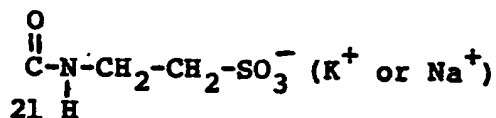
It is desirable that conjugated steroids have strongly ionized conjugating groups which are capable of forming micelles at low pH and concentrations (the critical micellar concentration, CMC, is a measure of this latter property). As mentioned above, examples of such desirable conjugating groups include but are not limited to taurine, homotaurine, sarcosyltaurine, and sulfobetaine. Steroids conjugated with such groups also have the advantages of stability and ease of synthesis.

Conjugation has additional effects as well, which provide the opportunity to tailor the conjugated steroid to a given clinical situation. For example, if the steroid is to be used to transport a drug across a mucosal

membran , e.g. the nasal mucosa, relatively long (e.g. h motaurine), branched (e.g. sarcosyltaurine), bulky (e.g. glucuronic acid), and amphoteric (e.g. sulfobetaine) groups are desirable, since they may cause the steroid to be held in the nasal membrane somewhat longer than unconjugated steroids or steroids conjugated with smaller conjugating groups.

Conjugation also, in some instances, lowers the CMC, so that only a small amount of steroid need be used. Strong acidic conjugating groups render the steroid resistant to being taken out of solution by variations in pH, ionic strength, and by the presence of other ions (e.g. Ca^{++}) and other macromolecules. Conjugation further prevents retention by the body, promotes rapid excretion, and prevents hepatic metabolism to potentially toxic metabolites (Beauaboin et al., J. Clin. Invest. (1975) 56:1431-1441).

As mentioned above, conjugating groups are bonded to C_{21} via any suitable linkage, e.g. amide or ester. Conjugation is carried out using conventional techniques. As an example, taurine bonded to C_{21} via an amide linkage is shown below (the presence of a cation, e.g. K^+ or Na^+ , is indicated in the parentheses):



The properties of the steroid are also affected by the nature of the substituents at C_3 , C_6 and C_{16} . Generally, OAc groups at these positions tend to aid solubility; however, OAc groups are also quite labile, and tend to decrease stability and shelf-life.

The fusidic acid or cephalosporin P_1-P_5 derivatives can be made by appropriately modifying commercially available fusidic acid or cephalosporin P_1-P_5 . Such techniques are known in the art.

The drugs which are admixed with a steroid carrier preferably have a molecular weight of between about 100 daltons and about 300,000 daltons. The drug may be either water soluble or lipid soluble, and may be a peptide, e.g. a peptide hormone such as insulin or a peptide hormone precursor such as proinsulin. Water soluble drugs, e.g. some peptides and vitamins, can also be transported across mucosal membranes by any of the steroids of the invention, including those whose unconjugated derivatives have relatively low k' values (between about 7 and about 15). For hydrophobic, lipid-soluble drugs, e.g. the lipid-soluble vitamins, the unconjugated derivative of the steroid should have a higher k' value, preferably above about 20.

Drugs for which the method of administration of the invention is particularly important are peptides. Suitable peptides include but are not limited to insulin, proinsulin, glucagon, parathyroid hormone and antagonists of it, calcitonin, vasopressin, renin, prolactin, growth hormone, thyroid stimulating hormone, corticotropin, follicle stimulating hormone, luteinizing hormone, chorionic gonadotropin, atrial peptides (a natriuretic factor), interferon, tissue plasminogen activator, gammaglobulin, Factor VIII, and chemical modifications of these peptides.

The invention can also be used to administer hormone releasing hormones, e.g. growth hormone releasing hormone, corticotropin releasing factor, luteinizing hormone releasing hormone, growth hormone releasing

inhibiting hormone (somatostatin) and thyrotropin releasing hormone.

5 Other suitable drugs include the physiologically
active enzymes: transferases, hydrolases, isomerases,
proteases, ligases, and oxidoreductases such as esterases,
phosphatases, glycosidases and peptidases; enzyme
10 inhibitors such as leupeptin, chymostatin and pepstatin;
and growth factors such as tumor angiogenesis factor,
epidermal growth factor, nerve growth factor and
insulin-like growth factors. Other suitable drugs are
those normally absorbed to a limited extent across the
gastrointestinal mucosa after oral administration; e.g.
15 antihistamines (e.g. diphenhydramine and
chlorpheniramine), and drugs affecting the cardiovascular
(e.g., antihypertensives), renal, hepatic and immune
systems (including vaccines). Additionally,
sympathomimetic drugs, such as the catecholamines (e.g.
20 epinephrine) and non-catecholamines (e.g. phenylephrine
and pseudoephedrine) may be administered according to the
method of the present invention.

 Drugs such as anti-infective agents, including
25 antibacterial, antiviral and antifungal agents may also be
administered according to the method of the present
invention. For example, antibiotics such as the
aminoglycosides (e.g., streptomycin, gentamicin,
kanamycin, etc.) are generally not adequately absorbed
30 after oral administration, and may therefore be
advantageously administered by the method of the invention.

 Many other drugs may also be administered
according to the invention, e.g. the many drugs currently
35 used to treat arthritis such as narcotic pain relievers.
Anti-inflammatory agents (e.g. indomethacin, dexamethasone
and triamcinolone), anti-tumor agents (e.g. 5-fluorouracil

and meth tr xat) and tranquiliz rs such as diaz pam may
als be administ r d acc rding t the invention.

5 Other suitable drugs are the water insoluble,
fat-soluble hydrophobic drugs, e.g. steroids such as
progesterone, estrogens (including contraceptives such as
ethinyl estradiol) and androgens and their analogs, and
the fat-soluble vitamins, e.g. vitamins A, D, E and K, and
10 their analogs.

The surface across which transport occurs may be
any mucosal surface such as the nasopharynx, conjunctiva,
oropharynx, ear canal, rectal, intestinal (enteral),
15 respiratory tract, endometrial, cervical vaginal, urethra,
urinary bladder or, in some circumstances, a skin surface
such as the axilla, the gluteal cleft, tympanic membrane,
between the toes, and the groin. Additionally, transport
of the drug according to the method of the present
20 invention may enhance penetration into the skin for
increased local effects.

The ratio of drug and steroid present in a
therapeutic composition will vary depending on a number of
25 factors, including the k' and CMC of the steroid, the
dosage of the drug to be administered, and the chemical
characteristics, e.g. hydrophobicity, of the drug.
Generally, the steroid is provided in an aqueous
physiological buffer solution which is then mixed with the
30 drug. The solution generally contains about 0.1% to about
2.5%, w/v, steroid in a physiologically acceptable
carrier, e.g. sodium phosphate buffered NaCl, pH 5-8,
having a NaCl concentration of about 0.05 M to about 0.6 M.

35 The concentration of the drug in the solution
will of c urs vary widely, depending n the nature of the

drug, and in the extent to which absorption is facilitated by the steroid. In some cases, administration according to the invention will enable the delivery of a higher dosage of the drug where needed than if the conventional mode of administration is used; in other cases, a much smaller dosage can be used because of efficient administration to a site. For instance, the amount of drug potentially can be decreased to one thousandth the amount or increased to ten times the amount of the drug normally used with conventional administration methods.

The therapeutic composition may contain, in addition to steroid and drug, any other desired non-toxic, pharmaceutically acceptable substances, e.g. a preservative such as phenol or cresol or stabilizing agents.

The dosage given at any one time will depend on a number of factors including, in addition to those mentioned above, the frequency of administration.

The compositions of the present invention may be administered to human and animal body surfaces in a variety of forms, including but not limited to, sprays, drops, suppositories, douches, salves, ointments, and creams. Some compositions may be advantageously applied in long term release dosage forms such as slow release, continuous release and intermittent release dosage forms. These long term release dosage forms include but are not limited to polymers, microcapsules, microspheres, osmotic diffusion devices and membrane release devices.

EXAMPLES

The following examples are intended to illustrate the invention, without acting as a limitation upon its scope.

Examples 1-14 demonstrate the effectiveness of using various fusidic acid derivatives (Le Pharmaceuticals, Ballerup, Denmark) and cephalosporin as adjuvants for the delivery of insulin, glucagon, human chorionic gonadotropin (hCG), proinsulin, corticotropin releasing-factor (CRF) and epinephrine across nasal mucosal membranes or conjunctival membranes in humans or sheep. Assays of insulin, glucagon, and hCG across nasal mucosal membranes and conjunctival membranes were accomplished by highly specific radioimmunoassay (RIA). (Protocols for RIA of insulin followed the procedures given in "GammaCoat [¹²⁵I] Insulin Radioimmunoassay Kit," Cat. No. CA-532, Clinical Assays Division of Travenol Laboratories, Inc., Cambridge, MA and for glucagon, the procedures given in "Protocol for the Radioimmunoassay of Glucagon [¹²⁵I]," Cat. No. 520, Cambridge Medical Diagnostics, Middle Billerica, MA) In the case of proinsulin, an insulin immunoassay with significant cross-reactivity for proinsulin was adapted to estimate the amounts of proinsulin absorbed. (The protocol is given in "Insulin [¹²⁵I] Radioimmunoassay," Corning Medical and Scientific, Medfield, MA.) The values obtained for insulin were adjusted to correlate with known cross-reactivity (i.e., 36%) of proinsulin in the assay. The "GammaDab [¹²⁵I] β -HCG Radioimmunoassay Kit," Clinical Assays Cat. No. 589, Clinical Assays Division of Travenol Laboratories, Inc., Cambridge, MA was used to evaluate hCG levels in serum.

INTRANASAL ADMINISTRATION OF INSULIN TO HUMANS

Example 1

The final concentration of insulin varied for each subject since they were of different weights.

Sodium tauro-24,25-dihydr fusidat was dissolved in 0.15 M NaCl, pH 7.4 to form a 5% solution, w/v. Commercially available porcine regular insulin (U-500) (Eli Lilly & Co., Indianapolis, IN) was mixed in a total volume of 2.0 ml with 0.15 M NaCl, pH 7.4 and the 5% solution of sodium tauro-24,25-dihydrofusidate to give final concentrations of 216 U/ml insulin and 1% (w/v) sodium tauro-24,25-dihydrofusidate. A normal human subject (subject 40) was administered at time 0, by nasal spray, two 75 microliter aliquots, so that the dosage of insulin administered to the subject was 0.5 Units/kg body weight.

For subjects 92 and 93, sodium tauro-24,25-dihydrofusidate was dissolved in 0.15 M NaCl, 0.05 M sodium phosphate buffer, pH 7.6 to form a 2% solution, w/v. Insulin was mixed in a total volume of 7.0 ml with the 0.15 M NaCl, 0.05 M sodium phosphate buffer and the 2% solution of sodium tauro-24,25-dihydrofusidate to give final concentrations of 220 Units/ml for subject 92 and 233 Units/ml for subject 93.

As shown in Table I, below, five minutes after nasal administration, subject 40's serum insulin level had increased more than twenty-fold, demonstrating that the insulin had been rapidly and effectively absorbed across the nasal mucosa. After ten minutes serum insulin levels for subjects 92 and 93 had increased more than ten-fold, indicating rapid absorption across the nasal mucosa. Furthermore, as is shown in Table I, each subjects' blood glucose was lowered significantly after twenty minutes, and was reduced more than fifty percent after thirty minutes for subjects 40 and 92.

TABLE I

5

STUDY OF NASALLY ADMINISTERED INSULIN IN
SODIUM TAURO-24,25-DIHYDROFUSIDATE TO HUMANS

		<u>Time</u> <u>(minutes)</u>	<u>Blood</u> <u>Glucose</u> <u>(mg/dl)</u>	<u>Serum</u> <u>Insulin</u> <u>(μU/ml)</u>	<u>Insulin</u> <u>(μU/ml</u> <u>above basal)</u>
10					
	SUBJECT 40	-20	89.9	5.6	1.1
		-10	88		
		0	89	4.5	0
		+5		115	110.5
		10	82	100	95.5
		15		95	90.5
15		20	54.5	62	57.5
		25		30	
		30	31.5	22	17.5
		40	37.5	10	5.5
		50	50	6.4	1.9
		60	72	8.8	4.3
		75	85	5	
		90	89	3.5	
20					
	SUBJECT 92	-20	81	5.0	.3
		0	80	4.7	0
		+5	79.5	52	47.3
		10	74.5	68	63.3
		15	61.5	55	50.3
		20	47	35	30.3
25		30	34.5		
		40	53.5	6.4	1.7
		50	67	3.2	0
		60	71.5	2.8	0
	SUBJECT 93	-20	91.5	7.2	1.8
30		0	89	5.4	0
		+5	89.5	54	48.6
		10	85	58	52.6
		15	75.5	28.5	23.1
		20	66	19.5	14.1
		30	65.5	7.2	1.8
		40	75	4.7	0
		50	84.5	9.4	4.0
35		60	84.5	6.2	0.8

Examp1 2

Unconjugated sodium fusidate was dissolved in
 5 0.15 M NaCl, pH 7.4 to form a 3% solution w/v.
 Commercially available insulin (U-500) was mixed in a
 total volume of 3 ml with 0.15 M NaCl, pH 7.4 and the 3%
 solution of sodium fusidate to give final concentrations
 of 210 Units/ml insulin and 1% (w/v) sodium fusidate. A
 10 normal human subject was administered the insulin
 preparation by nasal spray at time 0 as described in
 Example 1. The results obtained are shown in Table II
 below. Twenty minutes after nasal administration the
 subject's serum insulin level had increased more than
 fifteen-fold. Additionally, as is shown in Table II, the
 subject's blood glucose was reduced more than fifty
 percent after thirty minutes.

20

TABLE II

STUDY OF NASALLY ADMINISTERED INSULIN IN
UNCONJUGATED SODIUM FUSIDATE, pH 7.4 TO HUMANS

25

	<u>Time</u> <u>(minutes)</u>	<u>Blood</u> <u>Glucose</u> <u>(mg/dl)</u>	<u>Serum</u> <u>Insulin</u> <u>(μU/ml)</u>	<u>Insulin</u> <u>(μU/ml</u> <u>above basal)</u>
30	SUBJECT 33 -20	75.5	3.7	-.8
	0	79	4.5	0
	+5		16	11.5
	10	77	60	55.5
	15		80	75.5
	20	59.5	85	80.5
	30	34	28	23.5
	40	43.5	10.5	6
	50	56.5	7	2.5
	60	68.5	4.5	0

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Example 3

5 The insulin preparation was obtained by the same
procedure as described in Example 2. However, the pH of
the solution was adjusted to pH 7.95 and the final insulin
concentration was 267 Units/ml. A normal human subject
was administered the insulin preparation by nasal spray at
time 0 as described in Example 1. As shown in Table III
10 below the subject's serum insulin level had increased more
than thirty-five fold after fifteen minutes. Furthermore
the patient's blood glucose was lowered significantly
after thirty minutes.

15

TABLE III

20 STUDY OF NASALLY ADMINISTERED INSULIN IN
UNCONJUGATED SODIUM FUSIDATE, pH 7.95 TO HUMANS

20

	<u>Time</u> <u>(minutes)</u>	<u>Blood</u> <u>Glucose</u> <u>(mg/dl)</u>	<u>Serum</u> <u>Insulin</u> <u>(μU/ml)</u>	<u>Insulin</u> <u>(μU/ml</u> <u>above basal)</u>
SUBJECT 35	-20		1.7	0
25	0	80	1.7	0
	+5		45	43.3
	10	80	60	58.3
	15		34	32.3
	20	64	20	18.3
	30	46	10	8.3
	40	56	4.5	2.8
	50	73.5	14	12.3
30	60	84	7	5.3

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Example 4

Sodium glyco-24,25-dihydrofusidate was dissolved in 0.15 M NaCl, pH 7.6 to form a 3% solution w/v. Commercially available porcine insulin (U-500) was mixed in a total volume of 3.0 ml with 0.15 M NaCl, pH 7.6 and the 3% solution of sodium glyco-24,25-dihydrofusidate to give final concentrations of 216 Units/ml insulin and 1% (w/v) sodium glyco-24,25-dihydrofusidate. Subject 40 was administered the insulin preparation by nasal spray at time 0 as described in Example 1. As shown in Table IV below, five minutes after nasal administration, subject 40's serum insulin had increased more than twenty-fold, indicating that the insulin had been rapidly and effectively absorbed through the nasal mucosa. Furthermore, the subject's blood glucose was lowered significantly after twenty minutes.

TABLE IV

STUDY OF NASALLY ADMINISTERED INSULIN IN
SODIUM GLYCO-24,25-DIHYDROFUSIDATE TO HUMANS

	<u>Time (minutes)</u>	<u>Blood Glucose (mg/dl)</u>	<u>Serum Insulin (μU/ml)</u>	<u>Insulin (μU/ml above basal)</u>
SUBJECT 40	-20	90	3.5	.8
	0	86.5	2.7	0
	+5		66	63.3
	10	81	58	55.3
	15		38	35.3
	20	60	23	20.3
	30	51	6	3.3
	40	67.5	4.5	1.8
	50	78.5	5.6	2.9
	60	82.5	2.4	-.3

Example 5

For subject 34, sodium 24,25-dihydrofusidate was
5 dissolved in 0.15 M NaCl, pH 8.1 to form a 3% solution
w/v. Commercially available porcine insulin (U-500) was
mixed in a total volume of 3 ml with 0.15 M NaCl, pH 8.1,
and the 3% solution of sodium 24,25-dihydrofusidate to
10 give final concentrations of 190 Units/ml insulin and 1%
(w/v) sodium 24,25-dihydrofusidate. For subject 35, the
insulin preparation was made in the same manner as
described above except that the initial sodium
24,25-dihydrofusidate concentration was 3.75% (w/v) and
the final insulin concentration was 267 Units/ml. The two
15 normal human subjects were administered the insulin
preparation by nasal spray at time 0 as described in
Example 1. As shown in Table V below, fifteen minutes
after nasal administration subject 34's serum insulin
level had increased thirty-fold and the subject's blood
20 glucose was lowered significantly after thirty minutes.
Subject 35 showed a dramatic increase in serum insulin
(63.5-fold) after ten minutes, demonstrating that the
insulin had been rapidly and effectively absorbed through
the nasal mucosa. Additionally, the subjects' blood
25 glucose was lowered more than fifty percent after thirty
minutes.

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TABLE V

STUDY OF NASALLY ADMINISTERED INSULIN
IN SODIUM 24,25-DIHYDROFUSIDATE TO HUMANS

	<u>Time</u> <u>(minutes)</u>	<u>Blood</u> <u>Glucose</u> <u>(mg/dl)</u>	<u>Serum</u> <u>Insulin</u> <u>(μU/ml)</u>	<u>Insulin</u> <u>(μU/ml</u> <u>above basal)</u>
5				
10	SUBJECT 34 -20	76	2.2	.3
	0	76.5	1.9	0
	+5		9.6	7.7
	10	79	36	34.1
	15		38	36.1
	20	65.5	28	26.1
15	30	49	9.6	7.7
	40	44.5	7	5.1
	50	52.5	3	1.1
	60	69	2	1
20	SUBJECT 35 -20	75.5	1.6	-1
	0	76.5	2.6	0
	+5		94	91.4
	10	72	165	162.4
	15		130	127.4
	20	42.5	98	95.4
	30	21.5	37	34.4
	40	29	8.4	5.8
	50	44	6.4	3.8
25	60	54.5	3.5	.9

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Example 6

Cephalosporin P₁ was dissolved in 0.15 M NaCl, pH 7.6 to form a 5% solution, w/v. Commercially available insulin was mixed with 0.15 M NaCl, pH 7.6 and the 5% solution of cephalosporin P₁ to give final concentrations of 220 Units/ml insulin and 1% (w/v) cephalosporin P₁. A human subject was administered, by nasal spray two 75 microliter aliquots (33 Units) at time 0 so that the dosage of insulin administered to the subject was 0.5 Units/kg body weight.

As shown in Table VI, the subject's blood glucose level decreased slightly. Serum insulin levels are not yet available.

TABLE VI

STUDY OF NASALLY ADMINISTERED INSULIN
IN CEPHALOSPORIN P₁ TO HUMANS

	<u>Time</u> <u>(minutes)</u>	<u>Blood</u> <u>Glucose</u> <u>(mg/dl)</u>
	-15	96
	0	97.5
	+5	92
	10	91
30	15	90.5
	20	86.5
	30	76.5
	45	82.5
	60	88.5

CONJUNCTIVAL ADMINISTRATION OF INSULIN TO SHEEP

Example 7

5 Sodium tauro-24,25-dihydrofusidate was dissolved
in 0.15 M NaCl, 0.05 M sodium phosphate buffer, pH 7.6 to
form a 2% solution, w/v. Porcine insulin (U-500) was
mixed in a total volume of 1 ml with 0.15 M NaCl, 0.05 M
10 sodium phosphate buffer, pH 7.6 and the 2% solution of
sodium tauro-24,25-dihydrofusidate to give final
concentrations of 97.5 Units/ml insulin and 1% (w/v)
sodium tauro-24,25-dihydrofusidate. Two sheep were
administered 350 mg ketamine (a general anesthetic)
15 intravenously and 1.0 Unit/kg body weight of insulin to
the conjunctival sac as drops at time 0. As shown in
Table VII below, five minutes after conjunctival
administration, the serum insulin level of both sheep had
increased greater than five fold. Unlike human subjects,
20 sheep do not respond to these increments in serum insulin
levels with a decrease in blood glucose concentrations.

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TABLE VII

5	<u>STUDY OF CONJUNCTIVAL ADMINISTRATION OF INSULIN IN SODIUM TAURO-24,25-DIHYDROFUSIDATE TO SHEEP</u>		
10		<u>Time (minutes)</u>	<u>Serum Insulin (μU/ml)</u>
	Sheep A	-15	24
		-5	30
		0	24.5
		+5	>200
		10	>200
		15	>200
15		20	190
		30	>200
		45	195
		60	125
		75	70
		90	180
20	Sheep B	-15	22
		-5	22
		0	42
		+5	>200
		10	>200
		15	49
		20	35
		30	58
25		45	160
		60	82

INTRANASAL ADMINISTRATION OF GLUCAGON TO SHEEP

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Example 8

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Sodium tauro-24,25-dihydrofusidate was dissolved in 0.15 M NaCl, 0.5 M sodium phosphate, pH 7.6. The dry powder of bovine glucagon (Eli Lilly & Co., Indianapolis, Ind.) was dissolved in the fusidate buffer to a final

5 concentration of 1 mg/ml, pH 7.6. A sheep was administered 350 mg ketamine intravenously and 200 μ l of the glucagon solution was administered as a liquid at time 0, by drops into each side of the nose of the sheep. The dose of glucagon was approximately 10 μ g/kg body weight.

10 Serum glucagon levels were determined as shown in Table VIII. Five minutes after intranasal administration, the glucagon level had increased more than one hundred-twenty fold demonstrating that the glucagon had been rapidly and effectively absorbed through the nasal mucosa.

15 Glucagon was also administered to the nasal mucosa without adjuvant. However, no direct control studies were performed since glucagon is relatively insoluble at neutral pH. Instead, studies were performed as with the adjuvant, at pH 2.7. 420 μ g glucagon was
20 dissolved in 0.002 N HCl to administer the same concentration of material as described above. 200 μ g of glucagon was administered to each side of the nose at time 0. Glucagon was absorbed across the nasal mucosa without adjuvant at pH 2.7. However, the percentage of
25 increment was not as great without adjuvant as with adjuvant.

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TABLE VIII

STUDY OF NASALLY ADMINISTERED GLUCAGON WITH OR
 WITHOUT SODIUM TAURO-24,25-DIHYDROFUSIDATE TO SHEEP

		<u>Time (minutes)</u>	<u>Glucagon (pg/ml)</u>
10	SHEEP 412 ^a	-15	16
		-5	16
		0	30
		+5	3800
		10	1150
		15	660
15		20	340
		30	110
		45	130
		60	80
		75	105
		90	54
20	SHEEP 412 ^b	-15	210
		-5	
		0	450
		+5	4900
		10	3300
		15	1700
		20	960
		30	680
25		45	580
		60	800

^a Sodium tauro-24,25-dihydrofusidate (pH 7.6)

^b No sodium tauro-24,25-dihydrofusidate
 (highly acidic pH of 2.7 was necessary to dissolve
 glucagon at sufficient concentration for
 administration)

CONJUNCTIVAL ADMINISTRATION OF GLUCAGON IN SHEEP

Example 9

5 Sodium tauro-24,25-dihydrofusidate was dissolved
in 0.15 M NaCl, 0.05 M sodium phosphate, pH 7.6. Bovine
glucagon was dissolved in the fusidate containing buffer
to a final concentration of 1 mg/ml, pH 7.6. A sheep was
10 administered 350 mg ketamine intravenously and a total of
400 µg of the glucagon was administered to the
conjunctivae in the presence (pH 7.6) and absence (pH 2.7)
of adjuvant at time 0 and 90, respectively. As described
in Table IX below, in the presence of the adjuvant,
15 glucagon was absorbed across the conjunctival mucosa.
Since glucagon is relatively insoluble at neutral pH, no
direct control studies were performed. However, in the
absence of the adjuvant, glucagon was dissolved in 0.002 N
HCl, pH 2.7, as described in Example 8 above. The studies
20 indicated that glucagon was not absorbed across the
conjunctival mucosa to a significant extent.

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TABLE IX

STUDY OF CONJUNCTIVAL ADMINISTRATION
OF GLUCAGON WITH OR WITHOUT SODIUM
TAURO-24,25-DIHYDROFUSIDATE IN SHEEP

5

	<u>Time (minutes)</u>	<u>Glucagon (pg/ml)</u>
	Sheep 196 ^a	
	-15	250
	-5	
10	0	180
	+5	3400
	10	1200
	15	800
	20	620
	30	250
	45	310
15	60	280
	75	420
	Sheep 196 ^b	
	90	240
	95	580
	100	640
	105	560
	110	360
20	120	410
	135	470
	150	440

25 ^a Sodium tauro-24,25-dihydrofusidate (pH 7.6)
^b No sodium tauro-24,25-dihydrofusidate
(highly acidic pH of 2.7 was necessary to dissolve
glucagon at sufficient concentration for
administration)

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NASAL ADMINISTRATION OF HUMAN
CHORIONIC GONADOTROPIN IN SHEEP

Example 10

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Sodium tauro-24,25-dihydrofusidate was dissolved
in 0.15 M NaCl, 0.05 M sodium phosphate, pH 7.6. Human
chorionic gonadotropin (hCG, a glycoprotein of MW 39,000)
(National Pituitary Agency, Baltimore, MD) was dissolved
10 at a concentration of 2 mg/ml in 0.15 M NaCl and 0.5 M
sodium phosphate. The hCG was mixed with the fusidate
containing buffer to a final concentration of 1%, pH 7.6.
350 mg of ketamine was administered intravenously and one
mg of hCG was administered at time 0 in the form of drops
15 (250 µl in each nostril) of a sheep. As shown in Table X
below, the data indicate that in the presence of adjuvant,
significant blood levels of hCG did not appear in either
animal until 20 to 30 minutes after administration.

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TABLE X

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STUDY OF NASALLY ADMINISTERED HUMAN
CHORIONIC GONADOTROPIN IN SODIUM
TAURO-24,25-DIHYDROFUSIDATE IN SHEEP

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	<u>Time</u> <u>(minutes)</u>	<u>hCG</u> <u>(mIU/mL)</u>
SHEEP 412	-15	1.0
	-5	0
	0	1.5
	+5	2.5
	10	4.3
	15	6.4
	20	10.5
	30	18.0
	45	19.0
	60	28.0
	75	32.0
SHEEP 196	-15	0
	5	0
	0	0
	+5	0
	10	0
	15	0
	20	0
	30	4.1
	45	12.5
	60	15.5
	75	15.5
	90	19.5

INTRANASAL ADMINISTRATION OF PROINSULIN TO SHEEP

Example 11

5 Sodium tauro-24,25-dihydrofusidate was dissolved
in 0.15 M NaCl, pH 7.6. Commercially available proinsulin
(Eli Lilly & Co., Indianapolis, Ind.) was mixed with the
sodium tauro-24,25-dihydrofusidate solution or in 0.15 M
10 NaCl, pH 7.6 to give a final concentration of 6 mg/ml of
proinsulin and 1% (w/v) sodium tauro-24,25-
dihydrofusidate. Sheep were administered by nasal drops
1.5 mg (250 µl) of the solution into each nostril at
time 0. Sheep 25 was administered proinsulin without
15 adjuvant as described above. As shown in Table XI below,
proinsulin was rapidly absorbed in the presence of
adjuvant. However, in the absence of adjuvant the
proinsulin was absorbed at a slower rate and to a much
reduced extent.

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TABLE XI

5	STUDY OF NASALLY ADMINISTERED PROINSULIN WITH OR WITHOUT SODIUM TAURO-24,25-DIHYDROFUSIDATE TO SHEEP		
10		Time (minutes)	Proinsulin (μ U/ml above baseline)
15	SHEEP 408 ^a Study 24	-15	0
		0	0
		+5	119
		10	136
		15	133
		30	108
		60	53
		90	36
		120	33
20	SHEEP 196 ^b Study 25	180	53
		240	22
		-15	0
		0	0
		+5	0
		10	2.8
		15	28
		20	56
		30	22
25		60	0
		90	0
		120	0
		180	0
		240	2.8
30	^a Sodium tauro-24,25-dihydrofusidate ^b No sodium tauro-24,25-dihydrofusidate		

CONJUNCTIVAL ADMINISTRATION OF PROINSULIN TO SHEEP

Example 12

5 Sodium tauro-24,25-dihydrofusidate was mixed with
proinsulin as described in Example 11. Sheep were
administered a total of approximately 2.1 mg of the
proinsulin solution to the conjunctiva at time 0. As
10 indicated in Table XII below, proinsulin was rapidly
absorbed across the conjunctival membranes in the presence
of sodium tauro-24,25-dihydrofusidate.

15

TABLE XII

STUDY OF CONJUNCTIVAL ADMINISTRATION OF PROINSULIN
IN SODIUM TAURO-24,25-DIHYDROFUSIDATE TO SHEEP

20

	<u>Time</u> <u>(minutes)</u>	<u>Proinsulin</u> <u>(μU/ml above baseline)</u>
25	SHEEP 31	
	-15	0
	0	0
	+5	117
	10	83
	15	67
	20	53
	30	47
	45	25
	60	17
	90	42

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Example 13

INTRANASAL ADMINISTRATION OF
CORTICOTROPIN RELEASING FACTOR

5

Sodium tauro-24,25-dihydrofusidate was dissolved
in 0.15 M NaCl, pH 7.6 to form a 2% solution, w/v.
Corticotropin releasing-factor (CRF, a hormone of MW
4,000) (Dr. George Chrousos, National Institute of Health,
10 Bethesda, MD) was dissolved in 0.15 NaCl and mixed with
the 2% solution of sodium tauro-24,25-dihydrofusidate.
CRF was also prepared without adjuvant. Two sheep were
administered 350 mg ketamine intravenously and a total of
460 µg of CRF as a liquid, by drops into each side of the
15 nose of the sheep at time 0. The dose of CRF was
approximately 10 µg/kg body weight. As shown in Table
XIII, serum CRF levels were determined either with or
without adjuvant. Five minutes after intranasal
administration with adjuvant, the CRF level had increased
20 more than 200-fold. In contrast, without adjuvant the CRF
level increased only slightly. The data indicate that CRF
had been rapidly and effectively absorbed across the nasal
mucosa.

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TABLE XIII

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STUDY OF NASALLY ADMINISTERED CORTICOTROPIN
 RELEASING-FACTOR WITH OR WITHOUT
 TAURO-24,25-DIHYDROFUSIDATE IN SHEEP

	<u>Time</u> <u>(minutes)</u>	<u>CRF</u> <u>(pg/ml)</u>
10	SHEEP 11 ^a	
	0	15-20
	+5	>200
	10	>200
	15	>200
	20	>200
	30	>200
15	45	>200
	60	>200
	75	>200
	90	>200
	120	>200
	125	187
	130	186
	135	161
20	150	167
	165	148
	180	124
	195	98
	210	102
	SHEEP 12 ^b	
	0	16-20
25	+5	33
	10	45
	15	50
	20	39
	30	36
	45	28
	60	26
	75	25
30	90	25
	120	24

^a Sodium tauro-24,25-dihydrofusidate

^b No sodium tauro-24,25-dihydrofusidate

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INTRANASAL ADMINISTRATION OF EPINEPHRINE TO SHEEP

Example 14

5 Sodium tauro-24,25-dihydrofusidate was dissolved
in 0.05 M sodium phosphate buffer, pH 7.6 to form a 3%
solution. Commercially available epinephrine in solution
(1:1000) (Elkins Sinn, Richmond, VA) was mixed with 0.1 ml
10 of the 3% solution of sodium tauro-24,25-dihydrofusidate.
A sheep was administered 350 mg ketamine intravenously and
150 µl of the epinephrine solution as drops to each
nostril at time 0. The sheep was also administered
epinephrine as described above without adjuvant at a later
15 time.

Epinephrine levels were measured following
extraction of plasma and running on HPLC with
electrochemical detection.

20 As shown in Table XIV below, in the presence of
adjuvant, epinephrine was absorbed across the nasal mucosa
to a greater extent than without adjuvant.

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TABLE XIV

5

STUDY OF NASAL ADMINISTRATION OF EPINEPHRINE WITH OR
WITHOUT SODIUM TAURO-24,25-DIHYDROFUSIDATE IN SHEEP

		<u>Time (minutes)</u>	<u>Epinephrine (pg/ml)</u>
10	SHEEP 412 ^a	0	33
		+5	86
		10	134
		15	65
		30	78
	SHEEP 412 ^b	0	102
15		+5	323
		10	190
		15	229
		30	284

^a No sodium tauro-24,25-dihydrofusidate
^b Sodium tauro-24,25-dihydrofusidate

It is apparent that many modifications and
variations of this invention as herein above set forth may
be made without departing from the spirit and scope
thereof. The specific embodiments described are given by
way of example only and the invention is limited only by
the terms of the appended claims.

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W claim:

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1. A composition useful for the prevention or treatment of a human or animal disorder or for the regulation of a human or animal physiological condition comprising, in admixture:

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(a) as an active ingredient, a biologically-effective amount of a drug specific for the particular disorder or condition; and

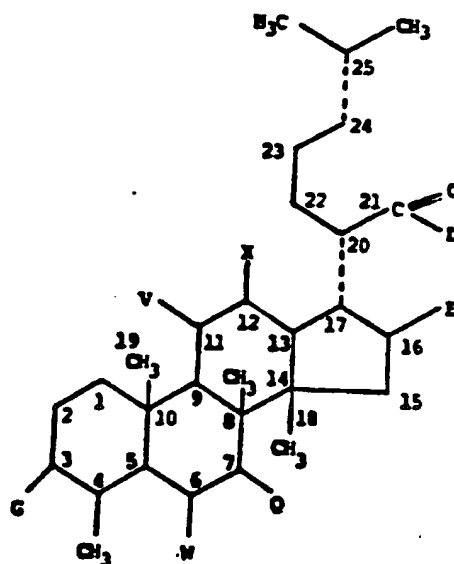
10

(b) as an adjuvant, a biocompatible, water-soluble, amphiphilic steroid, other than a natural bile salt, capable of increasing drug permeability of a human or animal body surface across which the drug is to be administered, in an amount effective to increase the permeability of said surface to said drug.

15

2. The composition of claim 1 wherein said amphiphilic steroid has the formula:

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wherein a dashed line represents a single or a double bond;

D represents a group having a molecular weight below about 600 daltons which renders an effective amount of said steroid water-soluble within a range of about pH 2 to about pH 12;

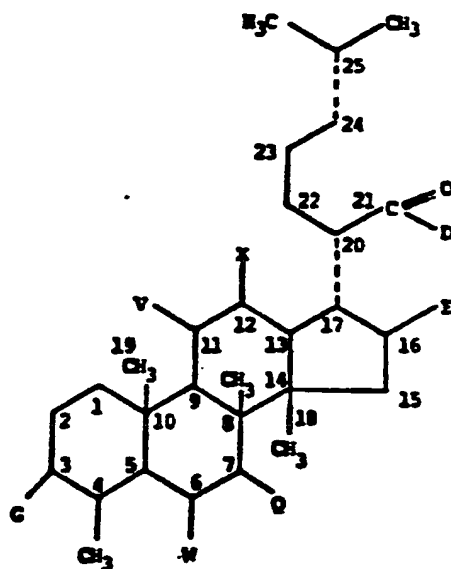
E and G each represent OAc, OH, a lower alkyl group or a lower heteroalkyl group;

W represents OAc or H; and

Q, V and X each represent H or OH,

said steroid containing from two to three polar functions, exclusive of the function represented by D.

3. The composition of claim 1 wherein said amphiphilic steroid has the formula:



wherein a dashed line represents a single or a double bond;

D represents a group having a molecular weight below about 600 daltons which renders an effective amount of said steroid water soluble within a range of about pH 2 to about pH 12;

E represents β OAc, α OH, a lower alkyl group in β position, or a lower heteroalkyl group in β position;

G represents α OAc, OH, a lower alkyl group, or a lower heteroalkyl group;

W represents α OAc or H;

Q represents H or OH, provided that, when W is α OAc and Q is α OH, Q is β -axial;

V represents H or α OH; and

X represents H or α OH,

said steroid containing from two to three OH groups.

4. The composition of claim 2 wherein the steroid is in unconjugated form, with D being selected from the group consisting of O^-Na^+ , O^-K^+ , O^-Rb^+ and O^-Cs^+ .

5. The composition of claim 2 wherein D is a covalently linked organic group which contains at least one carbon atom.

6. The composition of claim 5 wherein the covalently linked organic group is an amino acid containing an ionic function which is dissociated within the range of about pH 2 to about pH 12.

7. The composition of claim 6 wherein the amino acid is selected from the group consisting of glycine, taurine, homoglycine and homotaurine.

5

8. The composition of claim 6 wherein the amino acid is selected from the group consisting of sulfobetaine and phosphobetaine.

10

9. The composition of claim 5 wherein the covalently linked organic group is a peptide of two to three amino acids, said peptide containing an ionic function which is dissociated within the range of about pH 2 to about pH 12.

15

10. The composition of claim 9 wherein the peptide is selected from the group consisting of diglycine and glutathione.

20

11. The composition of claim 9 wherein the peptide is selected from the group consisting of sarcosylcysteine, hydroxyprolinetaurine, and sarcosyltaurine.

25

12. The composition of claim 5 wherein the covalently linked organic group is a heteroalkyl group of about three or fewer carbon atoms, said group containing an ionic function which is dissociated within the range of about pH 2 to about pH 12.

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13. The composition of claim 5 wherein the covalently linked organic group is a uronic acid of about six or fewer carbon atoms, said uronic acid containing an ionic function which is dissociated within the range of about pH 2 to about pH 12.

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14. The composition of claim 5 wherein the covalently linked organic group is a polyether containing between about six and about fourteen carbon atoms, inclusive, said polyether terminating in an ionic function which is dissociated within the range of about pH 2 to about pH 12.

15. The composition of claim 5 wherein the covalently linked organic group is a polyether containing between about sixteen and about twenty-four carbon atoms, inclusive.

16. The composition of claim 5 wherein the covalently linked organic group is a polyether containing between about sixteen and about twenty-four carbon atoms, inclusive, said polyether terminating in an ionic function which is dissociated within the range of about pH 2 to about pH 12.

17. The composition of claim 5 wherein the covalently linked organic group is bonded to C₂₁ of the steroid by an amide or an ester linkage.

18. The composition of claim 5 wherein the covalently linked organic group contains an ionic function, said ionic function being SO₃⁻, SO₄⁻, or COO⁻.

19. The composition of claim 3 wherein the steroid is further characterized in that

(a) its unconjugated derivative is retained on a column for a length of time sufficient to produce a k' factor value of at least about 4, said k' factor value being obtained by subjecting a monomeric solution of 1 mg/ml of such

unconjugated derivative to high-performance
(3,000 psi) liquid column chromatography using a
250 x 4.6 mm column having octadecylsilane-coated
5 μ m silica particles as the stationary phase and
a mobile phase, delivered at 1.0 ml/min.,
consisting of 75% methanol in water, v/v,
buffered with 0.005 M $\text{KH}_2\text{PO}_4/\text{H}_3\text{PO}_4$ to
give an apparent pH value, as measured using a
glass electrode, of pH 5.0, said k' factor value
being defined by $k' = \frac{t_r - t_0}{t_0}$,

where t_0 is the retention time in said column
of the solvent front and t_r is the retention
time in said column of said unconjugated
derivative as measured by obtaining the elution
profile of said steroid derivative by absorbance
at 210 nm;

(b) the critical micellar temperature of an
aqueous 1% solution, w/v, of the steroid is below
about 37°C within the range of about pH 2 to
about pH 12; and

(c) the critical micellar concentration of
the steroid is less than about 8 mMolar in 0.15 M
NaCl at 37°C, as measured by surface tension.

20. The composition of claim 19 wherein the
critical micellar temperature of the steroid is below
about 20°C and the critical micellar concentration is less
than about 4 mMolar.

21. A composition useful for the prevention or
treatment of a human or animal disorder or for the

regulation of a human or animal physiological condition comprising, in admixture:

5 (a) as an active ingredient, a
biologically-effective amount of a drug specific
for the disorder or condition; and

10 (b) as an adjuvant, an ionized or partially
ionized, water-soluble alkali salt of fusidic
acid or a derivative thereof, said fusidic acid
or derivative being capable of increasing drug
permeability of a human or animal body surface
across which the drug is to be administered, in
15 an amount effective to increase the permeability
of said surface to said drug.

22. A composition useful for the prevention or
treatment of a human or animal disorder or for the
regulation of a human or animal physiological condition
20 comprising, in admixture:

(a) as an active ingredient, a
biologically-effective amount of a drug specific
for the disorder or condition; and
25

(b) as an adjuvant, an ionized or partially
ionized, water-soluble alkali salt of
cephalosporin P₁, P₂, P₃, P₄ or P₅ or a
30 derivative thereof, said cephalosporin or
derivative being capable of increasing drug
permeability of a human or animal body surface
across which the drug is to be administered, in
an amount effective to increase the permeability
35 of said surface to said drug.

23. The composition of claim 21 wherein the derivative of fusidic acid is 24,25-dihydrofusidic acid.

5 24. The composition of claim 21 wherein the derivative of fusidic acid is 17,20-24,25-tetrahydrofusidic acid.

10 25. The composition of claim 21 wherein the derivative of fusidic acid is 3-acetoxyl-fusidic acid.

26. The composition of claim 21, 23, 24 or 25 wherein the fusidic acid or derivative thereof is conjugated at C₂₁.

15 27. The composition of claim 22 wherein the cephalosporin or derivative is conjugated at C₂₁.

20 28. The composition of claim 21 wherein the derivative of fusidic acid is tauro-24,25-dihydrofusidate.

29. The composition of claim 21 wherein the derivative of fusidic acid is tauro-16 α -OH-17,20-24,25-tetrahydrofusidate.

25 30. The composition of claim 21 wherein the derivative of fusidic acid is tauro-16 α -OH-24,25-dihydrofusidate.

30 31. The composition of claim 21 wherein the derivative of fusidic acid is tauro-17,20-24,25-tetrahydrofusidate.

35 32. The composition of claim 21 wherein the derivative of fusidic acid is glyco-24,25-dihydrofusidate.

33. The composition of claim 21 wherein the derivative of fusidic acid is tauro-16-O-methyl-ether-24,25-dihydrofusidate.

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34. The composition of claim 21 wherein the derivative of fusidic acid is tauro-16-O-methyl-ether-17,20-24,25-tetrahydrofusidate.

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35. The composition of claim 1, 21 or 22 wherein said drug is a peptide.

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36. The composition of claim 1, 21 or 22 wherein said drug is a peptide having a molecular weight between about 100 and about 300,000 daltons.

37. The composition of claim 35 wherein said peptide is a hormone or a precursor or inhibitor thereof.

20

38. The composition of claim 35 wherein said peptide is an enzyme or a precursor or an inhibitor thereof.

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39. The composition of claim 1, 21 or 22 wherein said drug is a glycoprotein.

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40. The composition of claim 1, 21 or 22 wherein said drug is selected from the group consisting of insulin, glucagon, parathyroid hormone, parathyroid hormone antagonist, calcitonin, vasopressin, renin, prolactin, growth hormone, thyroid stimulating hormone, corticotropin, corticotropin-releasing factor, follicle stimulating hormone, luteinizing hormone, chorionic gonadotropin, atrial peptides, interferon, tissue plasminogen activator, gammaglobulin, Factor VIII, growth hormone releasing hormone, luteinizing hormone releasing hormone and somatostatin.

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41. The composition of claim 1, 21 or 22 wherein said drug is an antihistamine.

5 42. The composition of claim 1, 21 or 22 wherein said drug is a sympathomimetic drug.

43. The composition of claim 1, 21 or 22 wherein said drug is an anti-infective agent.

10 44. The composition of claim 43 wherein said anti-infective agent is selected from the group consisting of antibacterial agents, antiviral agents and antifungal agents.

15 45. The composition of claim 1, 21 or 22 wherein said drug is an anti-arthritic agent.

20 46. The composition of claim 1, 21 or 22 wherein said drug is an anti-inflammatory agent.

47. The composition of claim 1, 21 or 22 wherein said drug is an anti-tumor agent.

25 48. The composition of claim 1, 21 or 22 wherein said drug is a tranquilizer.

49. The composition of claim 1, 21 or 22 wherein said drug is a cardiovascular agent.

30 50. The composition of claim 1, 21 or 22 wherein said drug is a water-insoluble, fat-soluble hydrophobic drug.

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5 51. The composition of claim 50 wherein the hydrophobic drug is selected from the group consisting of progesterone, an estrogen, ethinyl estradiol, an androgen, an androgen analog, vitamin A, vitamin D, vitamin E and vitamin K.

52. The composition of claim 1, 21 or 22 wherein said drug is a vaccinating agent.

10 53. The composition of claim 1, 21 or 22 wherein said drug is selected from the group consisting of renal and hepatic agents.

15 54. The composition of claim 37 wherein the hormone is insulin.

55. The composition of claim 37 wherein the hormone is glucagon.

20 56. The composition of claim 37 wherein the hormone is human chorionic gonadotropin.

25 57. The composition of claim 37 wherein the hormone is corticotropin-releasing factor.

58. The composition of claim 42 wherein the sympathomimetic drug is epinephrine.

30 59. A composition useful for the treatment of diabetes comprising, in admixture:

35 (a) as an active ingredient, a medically-effective amount of insulin or precursor thereof; and

(b) as an adjuvant, an ionized or partially ionized water-soluble alkali salt of fusidic acid or a derivative thereof, said fusidic acid or derivative being capable of increasing insulin permeability of a body surface across which the drug is to be administered, in an amount effective to increase the permeability of said surface to said drug.

60. The composition of claim 59 wherein the derivative of fusidic acid is a salt of tauro-24,25-dihydrofusidate.

61. The composition of claim 59 wherein the adjuvant is a salt of fusidic acid.

62. The composition of claim 59 wherein the derivative of fusidic acid is glyco-24,25-dihydrofusidate.

63. The composition of claim 59 wherein the derivative of fusidic acid is a salt of 24,25-dihydrofusidate.

64. The composition of claim 60, 61 or 63 wherein the salt is a sodium salt.

65. A composition useful for the treatment of diabetes comprising, in admixture:

(a) as an active ingredient, a medically-effective amount of insulin; and

(b) as an adjuvant, an effective amount of cephalosporin P₁.

66. A composition useful for regulating blood glucose and free fatty acid levels comprising, in admixture:

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(a) as an active ingredient, a medically-effective amount of glucagon; and

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(b) as an adjuvant, an effective amount of a salt of tauro-24,25-dihydrofusidate.

67. A composition useful for affecting human pregnancy comprising, in admixture:

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(a) as an active ingredient, a medically-effective amount of human chorionic gonadotropin; and

20

(b) as an adjuvant, an effective amount of a salt of tauro-24,25-dihydrofusidate.

68. A composition useful for regulating adrenocorticotropin synthesis and release comprising, in admixture:

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(a) as an active ingredient, a medically-effective amount of corticotropin-releasing hormone; and

30

(b) as an adjuvant, an effective amount of a salt of tauro-24,25-dihydrofusidate.

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69. A composition useful for the treatment of disorders or regulation of physiological conditions responsive to sympathomimetic drugs comprising, in admixture:

(a) as an active ingredient, a medically-effective amount of pinephrine; and

5 (b) as an adjuvant, an effective amount of a salt of tauro-24,25-dihydrofusidate.

70. The composition of claim 66, 67, 68 or 69 wherein the salt is a sodium salt.

10 71. A pharmaceutical preparation suitable for use as a nasal spray or nose drops comprising a solution or suspension of the composition of claim 1, 21, 22, 60, 61, 62, 63, 65, 66, 67, 68 or 69 in a physiologically acceptable carrier.

15 72. The pharmaceutical preparation of claim 71 wherein the physiologically acceptable carrier is a sodium phosphate buffer, a sodium phosphate buffered sodium chloride solution, or a sodium chloride solution.

20 73. A pharmaceutical preparation suitable for use as eye drops comprising a solution or suspension of the composition of claim 1, 21, 22, 60 or 66 in a physiologically acceptable carrier.

25 74. A pharmaceutical preparation in the form of a nasal spray or nose drops useful for the treatment of diabetes comprising a solution or suspension in a physiologically buffered sodium chloride solution of an admixture of a medically-effective amount of insulin as
30 active ingredient and an effective amount of sodium tauro-24,25-dihydrofusidate as adjuvant.

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